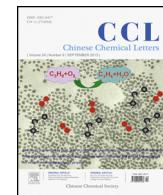




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## Original article

# A twin-axial[5]pseudorotaxane based on cucurbit[8]uril and $\alpha$ -cyclodextrin<sup>☆</sup>

Li-Hua Wang, Zhi-Jun Zhang, Heng-Yi Zhang, Hai-Lang Wu, Yu Liu \*

Department of Chemistry, State Key Laboratory of Elemento-Organic Chemistry, Nankai University, Tianjin 300071, China

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## ABSTRACT

A twin-axial hetero[5]pseudorotaxane was constructed based on 1-hexyl-4,4'-bipyridinium guest **1** and cucurbit[8]uril (CB[8]) and  $\alpha$ -cyclodextrin ( $\alpha$ -CD). In its structure, CB[8] included two bipyridinium units to realize the twin-axial mode, and the hexyl chain was threaded into the cavity of  $\alpha$ -CD. The [5]pseudorotaxane contains two types of macrocyclic hosts while the single axial and twin axial modes co-exist in its structure. The transformation of [5]pseudorotaxane could be realized by the addition of acid and 2,6-dihydroxynaphthalene (HN).

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## 1. Introduction

Various supramolecular assemblies, accompanied by novel structures or functions, have been constructed based on macrocyclic compounds [1–4]. As an important member of supramolecular assemblies, pseudorotaxanes have been used in construction of various molecular devices and machines because of their unique structures and stimuli-responsive properties [5,6].

The common [2]pseudorotaxane contains one thread-like component and one macrocyclic component [7]. It seems difficult to obtain the twin-axial pseudorotaxane which contains two guests threading into one macrocyclic component. Furthermore, the pseudorotaxane containing the twin-axial and single-axial mode simultaneously is very rare [8].

Crown ethers [9], cyclodextrins [10] and cucurbiturils [11,12] have been often used in the construction of pseudorotaxanes. To provide the system with more functions, we can introduce two or more kinds of cyclic components into one pseudorotaxane structure through accurate predesign. Both cyclodextrins and cucurbiturils are water soluble, but they exhibit different binding behaviors, which offers a good opportunity for introducing these two macrocycles into one system [13–15]. Cyclodextrins can include a neutral guest through hydrophobic interaction, which

have been applied in various supramolecular assemblies [16]. Cucurbiturils have also attracted great interest because of their unique binding ability toward cationic guests [17–24]. Among them,  $\alpha$ -cyclodextrin ( $\alpha$ -CD) shows a strong binding affinity toward an aliphatic chain, while cucurbit[8]uril (CB[8]) can accommodate more than one guest because of its large cavity.

Herein, we investigated the self-assembly behavior of 1-hexyl-4,4'-bipyridinium **1** and  $\alpha$ -CD and CB[8] (Fig. 1). A novel twin-axial hetero[5]pseudorotaxane was formed, in which CB[8] included two bipyridinium units and  $\alpha$ -CD included the hexyl group. Moreover, upon addition of acid and 2,6-dihydroxynaphthalene (HN), the structure of [5]pseudorotaxane can be transformed.

## 2. Experimental

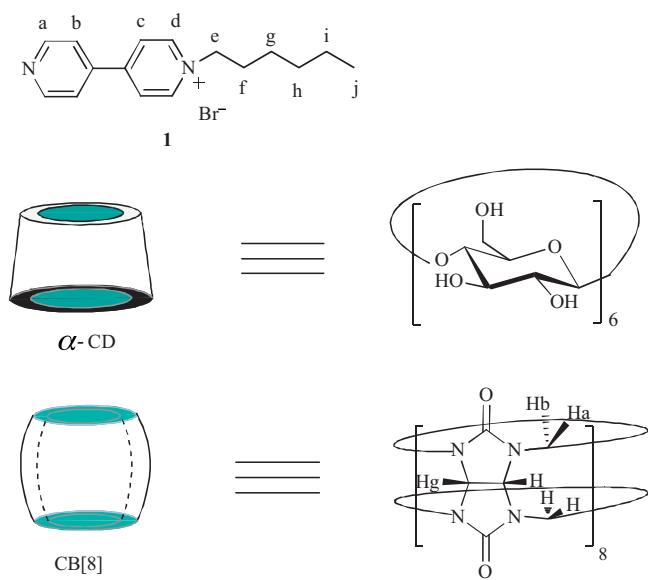
All chemicals were commercially available unless noted otherwise. NMR data were recorded on Bruker AV400 spectrometer, and chemical shifts were recorded in parts per million (ppm). Mass spectra were recorded using ThermoFinnigan LCQ Advantage ESI-MS.

*Preparation of compound **1*** [25]: 4,4-Dipyridyl (3.1 g, 20 mmol) and 1-bromohexane (5.5 g, 22 mmol) were dissolved in 35 mL acetonitrile, and refluxed for 18 h. After solvent was evaporated under vacuum, the residue was purified by column chromatography over silica gel (eluent: CHCl<sub>3</sub>/MeOH = 20/1) to afford **1** as a yellow solid (2.8 g, 44%). <sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O): δ 8.87 (d, 2H, J = 6.1 Hz), 8.69 (d, 2H, J = 4.7 Hz), 8.31 (d, 2H, J = 6.1 Hz), 7.82 (d, 2H, J = 4.7 Hz), 4.57 (t, 2H, J = 7.1 Hz), 1.97 (m, 2H), 1.25 (m, 6H), 0.77 (m, 3H).

\* Comment: A nice twin-axial hetero[5]pseudorotaxane constructed from CB[8] and  $\alpha$ -CD, could be transformed by the addition of acid and 2,6-dihydroxynaphthalene.

Corresponding author.

E-mail address: [yuliu@nankai.edu.cn](mailto:yuliu@nankai.edu.cn) (Y. Liu).

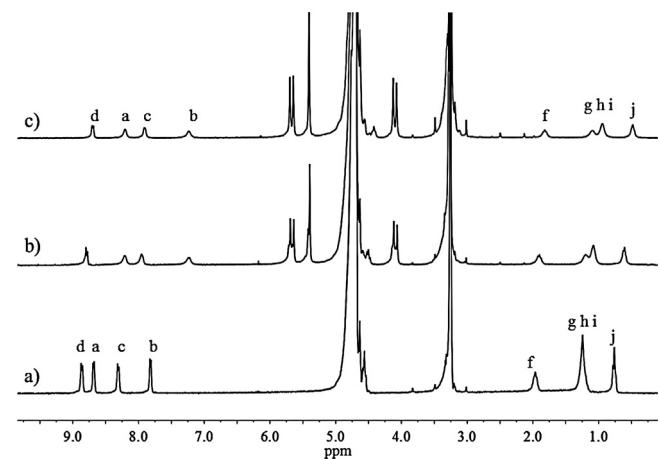
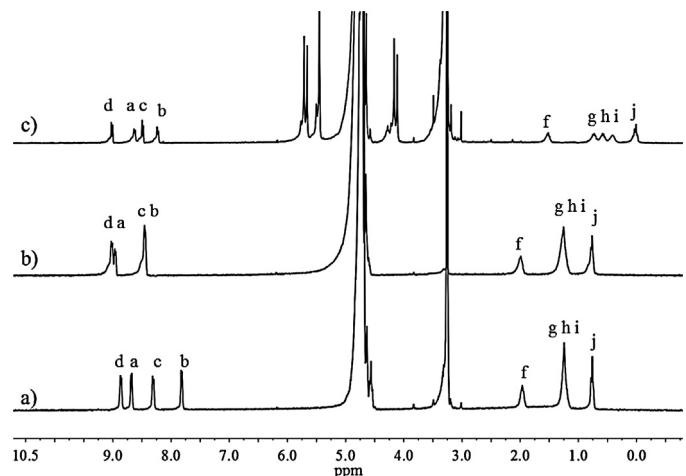
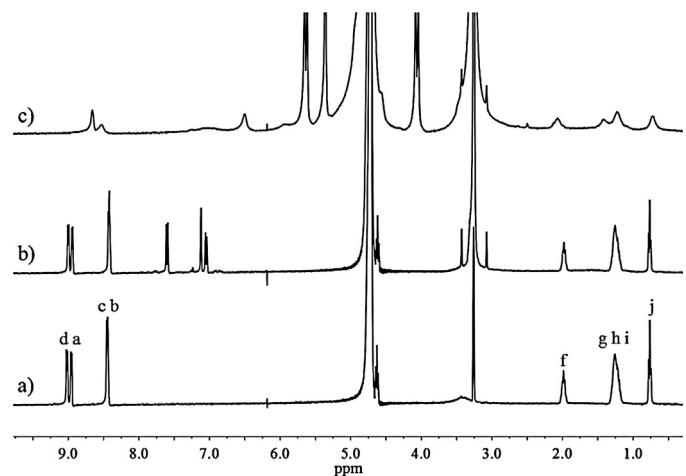
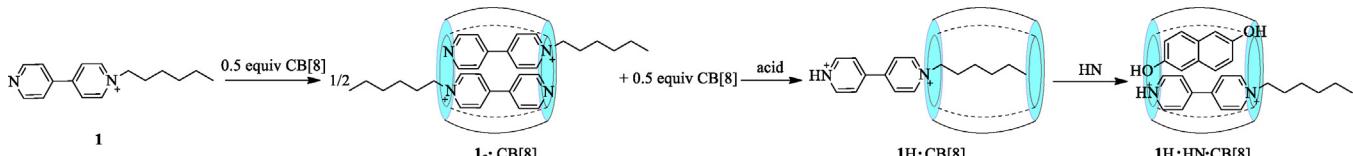
**Fig. 1.** The structures of guest **1** and hosts  $\alpha$ -CD.

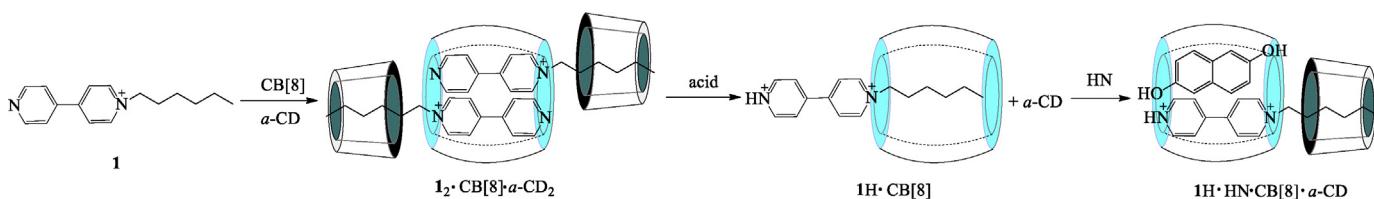
### 3. Results and discussion

#### 3.1. Self-assembly behavior of **1** and CB[8]

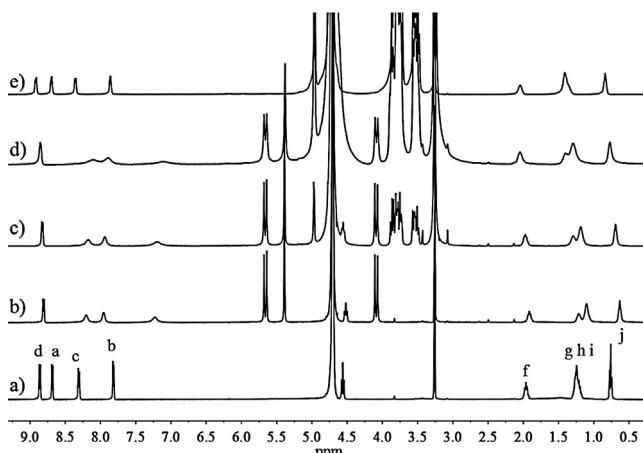
In our previous study, we found that the monocharged bipyridinium unit can form 2:1 complex with CB[8] [26,27]. But the complexation was strongly affected by the side chain. If the side chain possessed sufficient hydrophobicity, CB[8] would leave the bipyridinium unit and the twin-axial mode could not be maintained. In the structure of guest **1**, the side chain was a hexyl group (Fig. 2). We initially used  $^1\text{H}$  NMR to examine the self-assembly behavior of **1** and CB[8] (Fig. 3). After addition of 0.5 equiv. of CB[8], the signals of the aromatic protons H<sub>a</sub>, H<sub>c</sub> and H<sub>b</sub> all showed obvious upfield shifts, while the hexyl group only exhibited a minor shift. Further addition of 0.5 equiv. of CB[8] did not make any change in the signals of the guest, which suggested that a stable 2:1 complex was formed based on **1** and CB[8].

After **1** was acidified to **1H** (Fig. 4) and upon addition of 1.0 equiv. of CB[8], the proton signals of hexyl group were shifted upfield, while the aromatic signals only showed a minor shift, which suggested that the hexyl group was threaded into the cavity of CB[8]. All the observations indicated that before acidification, CB[8] preferred to include two bipyridinium units to maintain the twin-axial mode. After addition of acid, the monocharged bipyridinium unit acquired another positive charge at the bare N terminal and CB[8] moved to the hexyl group. The  $^1\text{H}$  NMR result (Fig. 5) showed that further addition of HN could attract the CB[8] to form a ternary charge transfer (CT) complex with the bipyridinium unit.

**Fig. 3.**  $^1\text{H}$  NMR spectra of **1** (1.0 mmol/L) upon addition of (a) 0 equiv., (b) 0.5 equiv., and (c) 1.0 equiv. of CB[8] ( $\text{D}_2\text{O}$ , 400 MHz, 298 K).**Fig. 4.**  $^1\text{H}$  NMR spectra of (a) **1** (1.0 mmol/L), (b) **1H** (1.0 mmol/L), (c) **1H** + 1.0 equiv. CB[8] (1.0 mmol/L) ( $\text{D}_2\text{O}$ , 400 MHz, 298 K).**Fig. 5.**  $^1\text{H}$  NMR spectra (a) **1H** (1.0 mmol/L), (b) **1H** + 1.0 equiv. HN (1.0 mmol/L), and (c) **1H** + 1.0 equiv. HN + 1.0 equiv. CB[8] (1.0 mmol/L) ( $\text{D}_2\text{O}$ , 400 MHz, 298 K).**Fig. 2.** Schematic representation of the binding modes and the interconversion process of pseudorotaxanes based on **1** and CB[8].



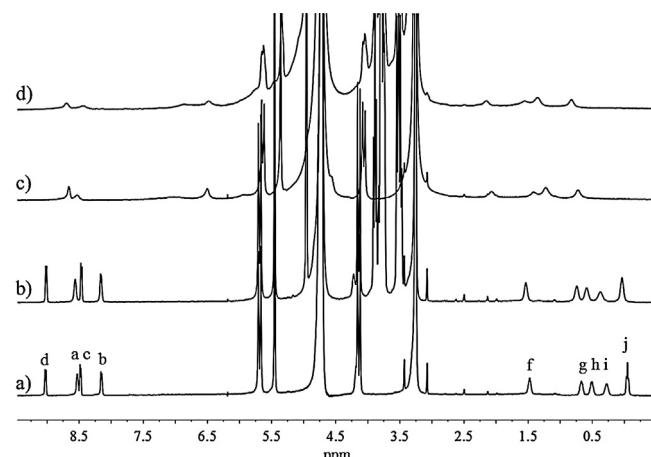
**Fig. 6.** Schematic representation of the binding modes and the interconversion process of pseudorotaxanes based on **1**,  $\text{CB}[8]$  and  $\alpha\text{-CD}$ .



**Fig. 7.**  $^1\text{H}$  NMR spectra (a) **1** (1.0 mmol/L), (b) **1** + 0.5 equiv.  $\text{CB}[8]$ , (c) **1** + 0.5 equiv.  $\text{CB}[8]$  + 1.0 equiv.  $\alpha\text{-CD}$ , (d) **1** + 0.5 equiv.  $\text{CB}[8]$  + 3.0 equiv.  $\alpha\text{-CD}$ , and (e) **1** + 3.0 equiv.  $\alpha\text{-CD}$  ( $\text{D}_2\text{O}$ , 400 MHz, 298 K).

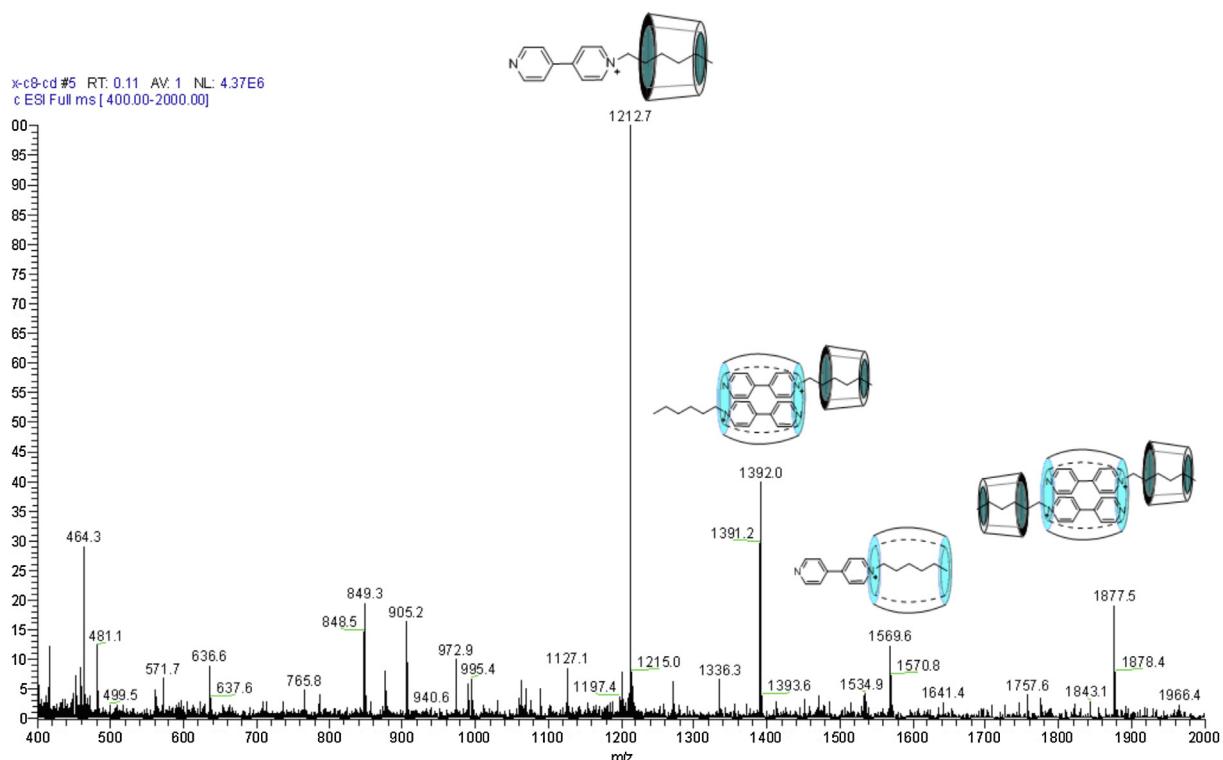
### 3.2. Self-assembly behavior of $\text{CB}[8]$ , $\alpha\text{-CD}$ and **1**

When  $\text{CB}[8]$  included the bipyridinium units, the unbound hexyl group became a binding site for  $\alpha\text{-CD}$  (Fig. 6). Upon



**Fig. 9.**  $^1\text{H}$  NMR spectra (a) **1H** + 1.0 equiv.  $\text{CB}[8]$  (1.0 mmol/L), (b) **1H** + 1.0 equiv.  $\text{CB}[8]$  + 3.0 equiv.  $\alpha\text{-CD}$ , (c) **1H** + 1.0 equiv.  $\text{CB}[8]$  + 1.0 equiv. HN, (d) **1H** + 1.0 equiv.  $\text{CB}[8]$  + 1.0 equiv. HN + 3.0 equiv.  $\alpha\text{-CD}$  ( $\text{D}_2\text{O}$ , 400 MHz, 298 K).

addition of 3.0 equiv. of  $\alpha\text{-CD}$  (Fig. 7d), the proton signals of the hexyl group shifted downfield, which was similar with the shift of **1**- $\alpha\text{-CD}$  (Fig. 7e). Meanwhile,  $\text{CB}[8]$  was still contained the aromatic units. These observations suggested that a [5]



**Fig. 8.** ESI-MS spectrum of the mixture of **1**,  $\text{CB}[8]$  and  $\alpha\text{-CD}$ . The peak at 1212.7 was assigned to  $[\mathbf{1}\cdot\alpha\text{-CD}]^+$ , 1392.0 was assigned to  $[\mathbf{1}_2\cdot\text{CB}[8]\cdot\alpha\text{-CD}]^{2+}$ , 1569.6 was assigned to  $[\mathbf{1}\cdot\text{CB}[8]]^+$ , and 1877.5 was assigned to  $[\mathbf{1}_2\cdot\text{CB}[8]\cdot\alpha\text{-CD}_2]^{2+}$ .

pseudorotaxane, **1<sub>2</sub>**·CB[8]· $\alpha$ -CD<sub>2</sub>, was formed, in which CB[8] encompassed two bipyridinium units to realize the twin-axial mode, and the hexyl chain of each molecule of **1** was threaded into the cavity of an  $\alpha$ -CD (Fig. 6). We further confirmed the formation of the complex by ESI-MS (Fig. 8). The peak at *m/z* 1212.7 was assigned to [1· $\alpha$ -CD]<sup>+</sup>, *m/z* 1392.0 was assigned to [1<sub>2</sub>·CB[8]· $\alpha$ -CD]<sup>2+</sup>, *m/z* 1569.6 was assigned to [1 CB[8]]<sup>+</sup>, and *m/z* 1877.5 was assigned to [1<sub>2</sub>·CB[8]· $\alpha$ -CD<sub>2</sub>]<sup>2+</sup>.

When we added  $\alpha$ -CD to the acidified solution containing **1H**·CB, the proton signals of guest showed only minor change (Fig. 9b) which indicated  $\alpha$ -CD could not drive CB[8] to disassociate the hexyl group. However, upon the addition of HN to **1H**·CB, we observed that the signals of aliphatic chain moved downfield. Further addition of  $\alpha$ -CD caused the signals of the aliphatic chain to exhibit a more downfield shift. This shift upon the addition of HN could be attributed to CB[8] moving to the bipyridinium unit to form a ternary CT complex, and the hexyl group threaded into  $\alpha$ -CD again to form the complex **1H**·HN·CB[8]· $\alpha$ -CD (Fig. 6).

#### 4. Conclusion

A novel twin-axial hetero[5]pseudorotaxane was constructed based on CB[8],  $\alpha$ -CD and guest **1**. In its structure, CB[8] included two bipyridinium units to maintain the twin-axial mode, and the hexyl chain was threaded into the cavity of  $\alpha$ -CD to realize the single-axial mode. Further addition of acid and HN could realize the transformation of [5]pseudorotaxane. The result presented here realized the co-existence of the single axial and twin axial modes in one pseudorotaxane structure, which will be beneficial for the construction of more complicated supramolecular assemblies with well-defined structures and functions.

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